

## COMPENSATION OF HUMAN VARIABILITY IN PULSE OXIMETRY

### Field of the Invention

5 The invention relates generally to pulse oximeters used to detect blood oxygenation. More specifically, the invention relates to a method for taking into account human variability in pulse oximeters. The invention further relates to a sensor allowing compensation for the inaccuracies caused by human variability, the sensor being an integral part of the pulse oximeter.

### 10 Background of the Invention

Pulse oximetry is at present the standard of care for continuous monitoring of arterial oxygen saturation (SpO<sub>2</sub>). Pulse oximeters provide instantaneous in-vivo measurements of arterial oxygenation, and thereby an early warning of arterial hypoxemia, for example.

15 A pulse oximeter comprises a computerized measuring unit and a probe attached to the patient, typically to a finger or ear lobe. The probe includes a light source for sending an optical signal through the tissue and a photo detector for receiving the signal after transmission through the tissue. On the basis of the transmitted and received signals, light absorption by the tissue  
20 can be determined. During each cardiac cycle, light absorption by the tissue varies cyclically. During the diastolic phase, absorption is caused by venous blood, tissue, bone, and pigments, whereas during the systolic phase there is an increase in absorption, which is caused by the influx of arterial blood into the tissue. Pulse oximeters focus the measurement on this arterial blood portion by  
25 determining the difference between the peak absorption during the systolic phase and the constant absorption during the diastolic phase. Pulse oximetry is thus based on the assumption that the pulsatile component of the absorption is due to arterial blood only.

Light transmission through an ideal absorbing sample is determined by  
30 the known Lambert-Beer equation as follows:

$$I_{out} = I_{in} e^{-\varepsilon DC}, \quad (1)$$

where  $I_{in}$  is the light intensity entering the sample,  $I_{out}$  is the light intensity received from the sample,  $D$  is the path length through the sample,  $\varepsilon$  is the extinction coefficient of the analyte in the sample at a specific wavelength,  
35 and  $C$  is the concentration of the analyte. When  $I_{in}$ ,  $D$ , and  $\varepsilon$  are known, and  $I_{out}$  is measured, the concentration  $C$  can be calculated.

In pulse oximetry, in order to distinguish between two species of hemoglobin, oxyhemoglobin ( $\text{HbO}_2$ ), and deoxyhemoglobin ( $\text{RHb}$ ), absorption must be measured at two different wavelengths, i.e. the probe includes two different light emitting diodes (LEDs). The wavelength values widely used are  
 5 660 nm (red) and 940 nm (infrared), since the said two species of hemoglobin have substantially different absorption values at these wavelengths. Each LED is illuminated in turn at a frequency which is typically several hundred Hz.

The accuracy of a pulse oximeter is affected by several factors. This is discussed briefly in the following.

10 Firstly, the dyshemoglobins which do not participate in oxygen transport, i.e. methemoglobin ( $\text{MetHb}$ ) and carboxyhemoglobin ( $\text{COHb}$ ), absorb light at the wavelengths used in the measurement. Pulse oximeters are set up to measure oxygen saturation on the assumption that the patient's blood composition is the same as that of a healthy, non-smoking individual.  
 15 Therefore, if these species of hemoglobin are present in higher concentrations than normal, a pulse oximeter may display erroneous data.

Secondly, intravenous dyes used for diagnostic purposes may cause considerable deviation in pulse oximeter readings. However, the effect of these dyes is short-lived since the liver purifies blood efficiently.

20 Thirdly, coatings like nail polish may in practice impair the accuracy of a pulse oximeter, even though the absorption caused by them is constant, not pulsatile, and thus in theory it should not have an effect on the accuracy.

Fourthly, the optical signal may be degraded by both noise and motion artifacts. One source of noise is the ambient light received by the  
 25 photodetector. Many solutions have been devised with the aim of minimizing or eliminating the effect of the movement of the patient on the signal, and the ability of a pulse oximeter to function correctly in the presence of patient motion depends on the design of the pulse oximeter. One way of canceling out the motion artefact is to use an extra wavelength for this purpose.

30 A further factor affecting the accuracy of a pulse oximeter is the method used to calibrate the pulse oximeter. Usually the calibration is based on extensive empirical studies in which an average calibration curve is determined based on a high number of persons. By means of this calibration curve, which relates the oxygen saturation of blood to pulse oximeter signals, the average  
 35 difference between the theory and practice (i.e. in-vivo measurements) is taken into account. The calibration curve typically maps the measured in-vivo signal to a corresponding  $\text{SpO}_2$  value.

Pulse oximeters, however, can also utilize the Lambert-Beer model for calculating the concentrations of the different Hb species. In this method of calibration, the measurement signals must first be transformed into signals applicable to the Lambert-Beer model for calculation. This transformation  
 5 constitutes the calibration of the pulse oximeter, since it is the step which adapts the in-vivo signals to the Lambert-Beer theory, according to which the pulse oximeter is designed to operate. Thus, the calibration curves can also be in the form of transformations used to adapt the actual in-vivo measurements to the Lambert-Beer model.

10 Transformations are discussed for example in U.S. Patent 6,104,938, which discloses a calibration method based on the absorption properties of each hemoglobin component, i.e. on the extinction coefficients of blood. In this method, the effective extinction coefficients are determined for each light signal via a mathematical transformation from the extinction coefficients according to  
 15 the Lambert-Beer theory.

However, each patient (i.e. subject of the measurement) has a calibration curve of his or her own, which deviates from the average calibration curve calculated on the basis of a high number of patients. This is due to the fact that for each patient the characteristics of the tissue through which light is  
 20 transmitted deviate from those of an average patient. One drawback of the current pulse oximeters is that they are incapable of taking this human variability into account. Human variability here refers to any and all factors causing patient-specific variation in the calibration curve, including time-dependent changes in the calibration curve of a single patient. As discussed in  
 25 the above-mentioned U.S. Patent, subject-dependent variation can also be seen as an effect of a third substance, such as a third hemoglobin species in the blood. However, the variation can also be interpreted as a subject-dependent change in the calibration curve of the pulse oximeter.

Without compensation for human variability, the accuracy of current  
 30 pulse oximeters is about  $\pm 2\%$  SpO<sub>2</sub>. However, in multi-wavelength applications in general, and especially if weak absorbers, such as COHb, are to be measured, the human variability represents a much more serious problem. Therefore, techniques of compensation for these inaccuracies are called for.

It is an objective of the invention to bring about a solution by means of  
 35 which the effects caused by the tissue of the subject can be taken into account when a pulse oximeter is calibrated. In other words, it is an objective of the present invention to create a pulse oximeter which can take into account the

differences caused by an individual subject as compared to the average calibration or transformation curve which the current pulse oximeter relies on.

A further objective of the invention is to bring about a general-purpose solution for the compensation of inaccuracies caused by human variability in pulse oximetry, a solution which is not limited to the particular general calibration method employed in the pulse oximeter, but which can be applied to any pulse oximeter regardless of its current built-in calibration method.

### Summary of the Invention

These and other objectives of the invention are accomplished in accordance with the principles of the present invention by providing a mechanism by means of which the subject-specific deviation in the tissue-induced effects on the accuracy of the pulse oximeter can be taken into account.

In the method of the invention, the effect of tissue is taken into account and the inaccuracies caused by subject-specific variation in that effect are compensated for. This is implemented by defining a nominal calibration for the apparatus and making off-line measurements, also termed "initial characterization measurements", in order to define the characteristics which describe the conditions under which the nominal calibration has been defined. Reference data indicating the characteristics are stored for subsequent on-line measurements in which light transmission through the actual tissue is measured. (Off-line here refers to measurements performed before the apparatus is taken into use, whereas on-line refers to the actual in-vivo measurements.) Subject-specific calibration is then defined based on the nominal calibration, the on-line measurements, and the reference data created in connection with the off-line measurements. Thus, the inaccuracies are eliminated by means of on-line measurements, which indicate the effect of the tissue. Off-line measurements are used to create the reference data so that light transmission measured subsequently through the tissue of a subject can be used to correct the nominal calibration for that particular subject.

Thus in one aspect the invention provides a method for compensating for subject-specific variability in an apparatus intended for non-invasively determining the amount of at least two light absorbing substances in the blood of a subject and being provided with emitter means for emitting radiation at a minimum of two different wavelengths and with detector means for receiving the radiation emitted, the method comprising the steps of

- calibrating the apparatus using a nominal calibration,
- carrying out initial characterization measurements, said measurements to include the measuring of radiation received by the detector,
- based on the characterization measurements, establishing nominal characteristics describing conditions under which the nominal calibration is used,
- storing reference data indicating the nominal characteristics established,
- performing in-vivo measurements on a living tissue, wherein radiation emitted through the tissue and received by the detector means is measured,
- based on the in-vivo measurements and the reference data stored, determining tissue-induced changes in the nominal characteristics, and
- compensating for subject-specific variation in the in-vivo measurements by correcting the nominal calibration on the basis of the tissue-induced changes.

In a preferred embodiment of the invention the method is divided in two steps so that the first step compensates for the inaccuracies caused by tissue-induced wavelength shift and the second step compensates for the inaccuracies caused by internal effects occurring in the tissue. The first step is then used to correct the extinction coefficients of the blood analytes to be measured, and the second step is used to correct the average transformation stored in the pulse oximeter.

In a further preferred embodiment of the invention the effect of the temperature is also compensated for in connection with the first step.

The method is not limited to pulse oximeters explicitly using the transformations, but can be applied to any pulse oximeter. However, the method is preferably applied to a pulse oximeter based on a transformation, since in a preferred embodiment the method is implemented by carrying out changes separately in the transformation and in the extinction coefficients.

In another aspect, the invention provides an apparatus for non-invasively determining the amount of at least two light absorbing substances in the blood of a subject, the apparatus comprising

- emitter means for emitting radiation at a minimum of two different wavelengths,
- detector means for receiving said radiation at each of said wavelengths and producing at least two electrical output signals,

- first signal processing means for processing said output signals and producing a modulation signal for each wavelength, whereby each modulation signal represents the pulsating absorption caused by the arterialized blood of the subject,

5        - second signal processing means for applying a predetermined calibration on said modulation signals, whereby transformed modulation signals applicable in the Lambert-Beer model are obtained,

- memory means for storing reference data indicating nominal characteristics under which said predetermined calibration has been applied,

10       - first compensation means, operatively connected to the memory means, for determining tissue-induced changes in the nominal characteristics,

- second compensation means, operatively connected to the first compensation means, for defining a subject-specific calibration by correcting the predetermined calibration on the basis of the tissue-induced changes, and

15       - calculation means, responsive to the second compensation means, for determining said amounts.

In a still further aspect, the invention provides a sensor for collecting measurement data for a pulse oximeter intended for non-invasively determining the amount of at least two light absorbing substances in the blood of a subject, the sensor comprising

20       - emitter means for emitting radiation at a minimum of two different wavelengths,

- detector means for receiving said radiation at each of said wavelengths and for producing at least two electrical output signals,

25       - storage means including reference data indicating nominal characteristics describing calibration conditions of the pulse oximeter, said data allowing apparatus connected to the sensor to determine tissue-induced changes in the nominal characteristics when radiation is emitted through said tissue.

30       Preferred embodiments of the invention are discussed in more detail below.

### **Brief Description of the Drawings**

35       In the following, the invention and its preferred embodiments are described more closely by referring to the examples shown in FIG. 1 to 12 in the appended drawings, wherein:

FIG. 1 illustrates the basic embodiment of a pulse oximeter according to the present invention,

FIG. 2 illustrates the signals utilized in the pulse oximeter of FIG. 1,

FIG. 3 shows the extinction coefficients of two different species of hemoglobin as a function of wavelength,

FIG. 4a to 4f illustrate the average transformation curves for two different pulse oximeters,

FIG. 5 is a flow diagram illustrating the prior art calibration method,

FIG. 6 is a flow diagram illustrating the general principle according to the present invention,

FIG. 7 is a flow diagram illustrating compensation for the inaccuracies caused by tissue-induced wavelength shift,

FIG. 8 illustrates an example of the transmission curve of human tissue, the curve being employed in the compensation of the inaccuracies caused by tissue-induced wavelength shift,

FIG. 9 illustrates the emitter characteristics determined for the compensation,

FIG. 10 is a flow diagram illustrating the compensation of the inaccuracies caused by internal effects occurring in the tissue,

FIG. 11 illustrates the low frequency baseline fluctuation utilized in a preferred embodiment of the invention, and

FIG. 12 illustrates an embodiment of a sensor according to the invention.

## **Detailed Description of the Invention**

Below, the solution according to the invention is discussed with reference to a pulse oximeter utilizing the above-mentioned transformations and four different wavelengths. As mentioned above, U.S. Patent 6,104,938 discloses a pulse oximeter utilizing the transformations.

FIG. 1 is a block diagram of a pulse oximeter utilizing four different wavelengths. Light from four different LEDs 10a, 10b, 10c, and 10d, each operating at a respective wavelength, passes into patient tissue, such as a finger 11. The light propagated through or reflected from the tissue is received by a photodetector 12, which converts the optical signal received into an electrical signal and feeds it to an input amplifier 13. The amplified signal is then supplied to a control unit 14, which carries out calculation of the amount of the Hb-derivatives in the blood. The control unit further controls the LED drive

**15** to alternately activate the LEDs. As mentioned above, each LED is typically illuminated several hundred times per second.

When each LED is illuminated at such a high rate as compared to the pulse rate of the patient, the control unit obtains a high number of samples at each wavelength for each cardiac cycle of the patient. The value of these samples (i.e. the amplitude of the received signal) varies according to the cardiac cycle of the patient, the variation being caused by the arterial blood, as mentioned above. The control unit **14** therefore utilizes four measurement signals, as shown in FIG. 2, each being received at one of the wavelengths.

**10** In order for variations in extrinsic factors, such as the brightness of the LEDs, sensitivity of the detector, or thickness of the finger, to have no effect on the measurement, each signal received is normalized by extracting the AC component oscillating at the cardiac rhythm of the patient, and then dividing the AC component by the DC component of the light transmission or reflection. The signal thus obtained is independent of the above-mentioned extrinsic factors. Thus in this case the control unit utilizes four normalized signals, which are in the following denoted with  $dA_i = \frac{AC_i}{DC_i}$ , where  $i$  is the wavelength in question (in

this basic embodiment of the multi-wavelength pulse oximeter  $i=1,2,3, 4$ ),  $AC_i$  is the AC component at wavelength  $i$ , and  $DC_i$  is the DC component at wavelength  $i$ . The signals  $dA_i$  are also referred to below as modulation signals. The modulation signals thus indicate how absorption is affected by the arterial blood of the patient.

The above-described measurement arrangement corresponds to a conventional four-wavelength pulse oximeter. The method of the present invention is implemented in the control unit of the pulse oximeter on the basis of the four modulation signals described above, i.e. the novelty of the system resides within the control unit itself. However, to be able to perform the self-calibration in conjunction with each patient, the control unit requires some pre-calculated data, which is stored in the memory (**M1**) of the pulse oximeter. Instead of being stored in conjunction with the control unit, this data, or at least part of it, can also be stored in the sensor part of the pulse oximeter. The sensor part, including at least the LEDs and the photo detector, is connected to the signal processing part, which includes the control unit. Consequently, depending on the overall configuration, the novelty can also reside partly in the sensor. The operation of the pulse oximeter is discussed in more detail below.

The theory of pulse oximetry is generally presented as being based on the Lambert–Beer Law. According to this theory, light transmission through the



tissue at each wavelength is exponentially dependent on the absorbance of the tissue (Eq. 1). This theory is generally accepted and established as a good model for pulse oximetry.

Next to be discussed is the theory and formalism on which the method of the invention is based.

According to the Lambert-Beer theory and for a system of two analytes, the signals described above can be presented as follows:

$$\begin{aligned}
 dA_1 &= dA \times (\varepsilon_1^{HbO_2} \times HbO_2 + \varepsilon_1^{RHb} \times RHb) \\
 dA_2 &= dA \times (\varepsilon_2^{HbO_2} \times HbO_2 + \varepsilon_2^{RHb} \times RHb) \\
 dA_3 &= dA \times (\varepsilon_3^{HbO_2} \times HbO_2 + \varepsilon_3^{RHb} \times RHb) \\
 dA_4 &= dA \times (\varepsilon_4^{HbO_2} \times HbO_2 + \varepsilon_4^{RHb} \times RHb) \\
 RHb &= 1 - HbO_2
 \end{aligned}$$

where  $dA$  is a common factor which depends on the absolute values, i.e. *inter alia* on the total amount of hemoglobin,  $\varepsilon_i^{HbO_2}$  is the extinction coefficient of oxyhemoglobin at wavelength  $i$  ( $i=1-4$ ),  $\varepsilon_i^{RHb}$  is the extinction coefficient of deoxyhemoglobin at wavelength  $i$ ,  $HbO_2$  is the concentration fraction of oxyhemoglobin, and  $RHb$  is the concentration fraction of deoxyhemoglobin.

Using a matrix notation, the above dependencies can be expressed for a system of  $n$  wavelengths and  $n$  analytes as follows:

$$\begin{pmatrix} dA_1 \\ dA_2 \\ \dots \\ dA_n \end{pmatrix} = C * \begin{pmatrix} \varepsilon_{11} & \dots & \varepsilon_{1n} \\ \varepsilon_{21} & \dots & \varepsilon_{2n} \\ \dots & \dots & \dots \\ \varepsilon_{n1} & \dots & \varepsilon_{nn} \end{pmatrix} \bullet \begin{pmatrix} HbX_1 \\ HbX_2 \\ \dots \\ HbX_n \end{pmatrix} \quad (2),$$

where  $dA_i$  is the differential change in absorption (i.e. the modulation signal) at wavelength  $\lambda_i$ ,  $\varepsilon_{ij}$  is the extinction coefficient of the hemoglobin derivative  $HbX_j$  at wavelength  $\lambda_i$ , and the constant  $C$  accounts for the change of units to fractional percentages of the concentrations of the analytes  $HbX_j$ .

FIG. 3 shows the extinction coefficients ( $\varepsilon^{HbO_2}$  and  $\varepsilon^{RHb}$ ) of oxyhemoglobin ( $HbO_2$ ) and deoxyhemoglobin ( $RHb$ ) as a function of the wavelength. Point **P** shown in the figure is the isobestic point of oxyhemoglobin ( $HbO_2$ ) and deoxyhemoglobin ( $RHb$ ). The point has the special property that the modulation signal at the wavelength in question does not depend on the respective proportions (relative concentrations) of the hemoglobin species. Thus at the wavelength of point **P** the effect of the relative concentrations of oxyhemoglobin and deoxyhemoglobin on the result of the measurement is nil. It should be noted, however, that the modulation signal is independent of the

relative concentrations only, not of the absolute concentrations. Thus, the absolute amount of the hemoglobin species has an effect on the result of the measurement.

As is known, there is a difference between the Lambert-Beer theory and the practical measurements. The difference is due to the fact that the Lambert-Beer theory does not take into account the scattering and non-homogeneity of the tissue, whereas the actual extinction coefficients are also dependent on the scattering of light caused by the tissue and blood, and on the combined effect of absorption and scattering. The larger the proportion of the attenuation caused by absorption and scattering, the larger is the correction needed between the actual and the theoretical (non-scatter) domains. This correction between these two domains can be represented by the transformation curves discussed above, by means of which the actual in-vivo measurements are mapped to the Lambert-Beer model.

The transformation can be expressed, for example, as follows:

$$N_{kl}^{L-B} = g_{kl}^{-1} (N_{kl}^{in-vivo}) \quad (3),$$

where  $N_{kl} = \frac{dA_k}{dA_l}$  is the modulation ratio (the superscript indicating the domain) in the form of a polynomial function (the subscripts  $k$  and  $l$  indicating the wavelengths in question), and  $g$  is the transformation (the subscript “-1” denoting the inverse function).

Figures 4a to 4f illustrate the average transformation curves measured for a pulse oximeter, where the two wavelengths for measuring the two species of hemoglobin are 660 nm and 900 nm and the third wavelength is either 725 nm or 805 nm. Figures 4a to 4c illustrate the transformation curves for a pulse oximeter with the third wavelength being 725 nm, and Figures 4d to 4f illustrate the transformation curves for a pulse oximeter with the third wavelength being 805 nm. Each curve shows the Lambert-Beer  $N_{k,l}$  as a function of the in-vivo  $N_{kl}$  at wavelengths  $k$  and  $l$ .

FIG. 5 is a flow diagram describing the general measurement principle described in U.S. Patent 6,104,938. In this method, the above-mentioned  $N_{kl}^{in-vivo}$  values are first determined from the  $dA_i$  values measured (step 51). The average transformations  $g_{kl}$  are then used to convert the measured in-vivo values to values  $N_{kl}^{L-B}$ , which can be used in the ideal Lambert-Beer model (step 52). Other input values needed for the Lambert-Beer model are also determined (step 53). In practice these input values are the ideal (nominal) extinction coefficients of the analytes to be measured, the extinction coefficients

being given for the center wavelengths used in the measurement. The converted transformation values and the nominal input values (i.e. nominal extinction coefficients) are then used according to the Lambert-Beer model to calculate the concentrations of the desired analytes (step 54). Thus in this approach the in-vivo values  $N_{kl}^{in-vivo}$  measured from the tissue are converted to the ideal in-vitro (cuvette) environment, where the ideal oximetry model (i.e. the Lambert-Beer model) is applied to yield the desired concentrations.

In the standard two wavelength pulse oximetry the prior art technique is to map the modulation ratio  $N_{kl}^{in-vivo}$  directly to the SpO2 percentage measured. In this simple case the transformation is not necessary, though the transformation technique together with the solution in the Lambert-Beer domain can be utilized as well.

There are two basic ways to determine the average transformation, a theoretical approach and an empirical approach. In the empirical approach the measurements are made in the tissue by taking blood samples and measuring the actual proportions of the hemoglobin species and then determining the value of  $N_{kl}^{L-B}$  on the basis of the measured proportions. The transformation is then obtained as the relationship between the values based on the blood samples and the values given by empirical measurements as measured by the pulse oximeter. The theoretical approach, in turn, is based on a known tissue model, which takes into account the characteristics of the tissue as referred to above, which are ignored in the Lambert-Beer model. A first value is determined for in-vivo  $N_{kl}$  by means of the tissue model and a second value on the basis of the Lambert-Beer model. The tissue parameters of the model are determined so that the known 2-wavelength calibration (so-called R-curve) is reproduced. Then using these tissue parameters and the wavelength dependence of the tissue model, the relation of the in-vivo  $N_{kl}$  and the Lambert-Beer  $N_{kl}$  is extrapolated to other wavelengths in order to obtain the transformations at these new wavelengths. Thus in the theoretical approach no new empirical measurements are made.

In practice the transformation can be a quadratic equation yielding a correction of the order of 20 percent to the measured  $N_{kl}^{in-vivo}$  value, for example. As discussed below, the transformation data (i.e. the transformation curves) are preferably stored in numeric form in the pulse oximeter or the sensor. The number of transformation curves stored in the pulse oximeter can vary, depending on the number of wavelengths used, for example. Typically there is a transformation curve for each wavelength pair.

As mentioned above, the accuracy of a pulse oximeter utilizing an average transformation is not necessarily sufficient, especially if analytes which are weak absorbers are to be measured or if two analytes absorb similarly, whereby it is difficult to distinguish the said analytes from each other.

5 In the present invention the accuracy of the pulse oximeter is improved by taking into account the subject-specific light transmission through the tissue, and changing the values input to the ideal model, i.e. the nominal transformation and the nominal extinction coefficients, on the basis of the measurement to compensate for the subject-specific changes.

10 FIG. 6 is a flow diagram illustrating the general principle of the present invention. In the method of the invention, initial characterization measurements are first made off-line, preferably at the calibration stage of the pulse oximeter with a nominal wavelength pulse oximeter sensor (step 61). Based on the measurements, nominal characteristics are established describing the  
15 conditions under which the pulse oximeter has been calibrated (steps 62). As a result of these steps, reference data are stored (step 63), which describe the calibration conditions of the pulse oximeter. In connection with subsequent in-vivo measurements, the same characteristics are again estimated and tissue-induced changes in the characteristics are determined based on the measured  
20 characteristics and the reference data stored (step 64). In in-vivo measurement, the  $N_{kl}^{in-vivo}$  values are determined from the  $dA_i$  values measured. On the basis of the changes determined, the subject-specific calibration is then determined (step 65) for the in-vivo measurements to be performed by the pulse oximeter on the subject.

25 It is to be noted here that steps 61-63 are off-line steps performed either when the pulse oximeter has been calibrated in a known manner or at the manufacturing stage of the pulse oximeter sensor when the sensor characteristics are determined. After these steps, the nominal transformation and the nominal extinction coefficients are known to the pulse oximeter.

30 Human tissue can influence the accuracy of a pulse oximeter by two different mechanisms. First a direct wavelength shift is caused in the LED emission due to the filtering effect of the tissue. Namely, on one side of the LED center wavelength the absorption may be larger than on the other, whereby the center wavelength is effectively shifted towards the region with  
35 smaller absorption. The second mechanism is a subtle one. It arises from the fact that the arterial blood is in interaction with the surrounding tissue, which can either increase or decrease the effective path length through the arterial

blood layer. The first mechanism is in this context termed the external mechanism, since it affects factors external to the tissue (wavelength). The second mechanism is called the internal mechanism, as it is caused by internal factors in the tissue itself.

5 Therefore, the subject-specific variations in the influence of tissue are preferably compensated for by adding two compensating processes to the above-described prior art mechanism; the first process attends to the subject-specific variation in the external mechanism, and the second one to the subject-specific variation in the internal mechanism. The first process preferably  
10 controls the extinction coefficients to be input to the Lambert-Beer model, while the second process preferably controls the value of the transformation used to transform the modulation ratios  $N_{kl}^{\text{in-vivo}}$  to the Lambert-Beer model  $N_{kl}^{\text{L-B}}$ . The linear equations with the unknown analyte concentrations are then solved in the Lambert-Beer model, as in the prior art method. The degree of these  
15 compensations is determined by DC light transmission through the tissue (the measured DC signal), measured both in off-line and on-line conditions.

The first process is illustrated in the flow diagram of FIG. 7. Initial off-line measurements are again made in order to determine the nominal characteristics under which the pulse oximeter has been calibrated (steps 71  
20 and 72). As a result of these steps, reference data is created and stored, the data indicating the nominal characteristics (step 73). On the basis of the data and the in-vivo measurements made, the tissue-induced changes in the extinction coefficients are then determined in connection with the in-vivo measurements (step 74). The new coefficients are then used in the Lambert-  
25 Beer model to solve the concentrations of the analytes (step 75). Thus in this embodiment the extinction coefficients are changed in a subject-specific manner in order to improve the accuracy of the pulse oximeter.

In one preferred embodiment of the invention the compensation for any other wavelength shift type interference than tissue-induced interference,  
30 such as a shift caused by high LED temperature, spectral absorption, or reflection properties of the sensor cushion or adhesive, is also implemented in connection with compensation for the filtering effect of the tissue.

In the following the compensations are discussed in more detail by disclosing a practical implementation for both compensation steps. The  
35 compensation of subject-variability causing wavelength shift type interference (i.e. external mechanism) is discussed first.

In the Lambert-Beer model (see Eq. 2) the effective extinctions  $\varepsilon_{ij}^{\text{effective}}$  for broadband emitters, such as LEDs, can be calculated as follows:

$$\varepsilon_{ij}^{\text{effective}} = \frac{1}{W} \int_{\Delta\lambda} \varepsilon_j(\lambda) * LED_i(\lambda(T)) * DET(\lambda) * tissue(\lambda) d\lambda \quad (4),$$

where the integration is over the LED emission spectrum  $LED_i(\lambda(T))$ ,  
 5  $DET(\lambda)$  represents the spectral sensitivity of the detector,  $tissue(\lambda)$  is the spectral transmission of light through the tissue,  $\varepsilon_j(\lambda)$  is the spectral extinction of the analyte in question,  $T$  is the temperature, and  $W = \int LED * DET * tissue * d\lambda$  represents a normalization factor.

In a preferred embodiment of the invention, the radiation emitting  
 10 means are Light Emitting Diodes (LED), but lasers emitting at one single wavelength are also possible. For lasers the effective extinction values are the extinction values at the laser wavelength, which can depend, however, on the temperature of the emitter component. In the case of a laser, Eq. 4 is thus not needed to calculate the effective extinction value. In the preferred embodiment  
 15 of the invention the emitter and detector means are located at the tissue site at which the radiation is transmitted through the tissue, but the radiation can also be conducted to and from the tissue site in a light conducting fiber or in equivalent conduction means. In this case Eq. 4 shall also include a term for the spectral transmission of the radiation conductor. A sensor utilizing light  
 20 conducting fibers can be as shown in FIG. 5 of the above-mentioned U.S. Patent 6,104,938.

The extinction coefficients can thus be calculated according to this equation by determining all the above factors, which depend on the actual wavelength values. However, as the task of determining the exact value of Eq.  
 25 4 is not possible in connection with a real-time pulse oximeter measurement using only a few discrete wavelength bands, in practice the result of Eq. 4 has to be approximated. The compensation is based on determining nominal extinction coefficients and approximating their wavelength dependence in advance at the factory and using this information in the real measurement  
 30 situation to approximate the final subject-specific extinction coefficients.

The compensation algorithm will now be presented for a 4-wavelength pulse oximeter according to FIG. 1, having four LEDs at nominal wavelengths of 627nm, 645nm, 670nm, and 870nm. The extinction matrix for RHb (first column), HbO<sub>2</sub>, HbCO, and metHb (last column) and for the above four  
 35 wavelengths (627nm on top) is then nominally in L/(mmol\*cm).

$$E_{kl}^0 = \begin{pmatrix} 1.132 & 0.1799 & 0.2734 & 3.575 \\ 0.9182 & 0.1124 & 0.1337 & 2.411 \\ 0.7353 & 0.0885 & 0.0550 & 0.5796 \\ 0.2071 & 0.2772 & 0.010 & 0.5754 \end{pmatrix} \quad (5)$$

The above extinction coefficients have been calculated applying Eq. 4 at nominal LED drive temperature without the tissue filtering term  $tissue(\lambda)$ . It then represents a nominal extinction matrix for a SpO2 sensor before its attachment on the tissue site. This extinction matrix is then altered on the basis of the measured filtering effect caused by the tissue, when the sensor is attached on the site.

We next introduce a parameter called the Functional Light Transmission (FLT)<sub>i</sub> at a wavelength  $i$ , since it is used below in order to make all DC<sub>i</sub> values (measured at varying LED emission powers at the four discrete wavelengths  $i$ ) comparable to each other. Using DC values comparable to each other is in practice a prerequisite for unveiling the real effect of the tissue on the measurements and the characteristics of the tissue. In order to obtain the comparable units, the DC light transmission for each LED channel (wavelength) is first measured at a certain emitter drive current, and the measured DC value is then reduced in the preamplifier to a detector current, which is normalized to an emitter current value of 1 mA. When measured without the tissue in the probe, this result is called the Current Transfer Ratio (CTR) of the probe. CTR characterizes the sensor design and the efficiency of the light transmission from the emitters to the detector. It is usually of the order of a few microAmps (of detector current) per one milliAmp (of LED current). Now the tissue (e.g. a finger) is inserted into the probe and the CTR is again measured. This result is now called the Functional Current Transfer Ratio (FCTR) because it is the CTR measured under conditions of the function of the pulse oximeter, i.e. when the tissue is in place in the probe. The FLT<sub>i</sub> is then calculated for each emitter (wavelength) as follows:

$$FLT(emitter\#k) = FCTR(1mA - emitter - current) / CTR(1mA - emitter - current).$$

Next the CTR and the FCTR concepts will be linked to the Lambert-Beer absorption model and to the actual measured intensities in the pulse oximeter. The CTR obviously describes how the external probe design factors, such as the color and geometry of the probe, affect the light transmission to the detector. On the other hand, the FLT can be associated with the true transmission through the tissue in units which are normalized to the emitter

efficiency. Therefore, Eq. 1 can be written in a slightly different form, as it is often written in transport theory:

$$I = I_0 \exp(-\alpha * d) = I_0 \exp(-\alpha_{int} * d) \exp(-\alpha_{ext} * d') \quad (6),$$

where  $d$  is the tissue thickness and  $\alpha$  is an effective absorption coefficient. The above equation has been divided into two components. The attenuation factor with  $\alpha_{int}$  accounts for all internal absorption effects, such as blood and can be associated to the FLT-value as the FLT equals one when no tissue (no internal attenuation) is in the probe, and the factor with  $\alpha_{ext}$  accounts for all external attenuations, such as geometrical factors and multiple surface reflections without light penetration into the tissue, and can be associated with the CTR of the probe. (The term  $d'$  denotes the 'phantom' absorption thickness parameter for the external effects.) The term  $\alpha_{ext}$  is mainly a SpO2 probe design issue which does not influence the measurement accuracy as such, and thus it need not to be compensated for by any means. The FLT at wavelength  $k$  can now be defined as:

$$FLT_k = \frac{I}{I_0 \exp(-\alpha_{ext} * d')} = \exp(-\alpha_{int} * d) = FCTR_k / CTR_k \quad (7).$$

The FLT thus describes light attenuation caused by the tissue, and it can be related to the DC light transmission in the pulse oximeter.

It is now assumed that the spectral tissue transmission is as presented in FIG. 8, which shows spectral characteristics of tissue in the same units for each wavelength, i.e. FLT as a function of the wavelength, based on an empirical measurement. In a continuous real-time SpO2 measurement, the transmission is known at 4 distinct wavelength values (the FLT values derived from the DC values in the pulse oximeter) marked in the figure. At each wavelength the slope of the tissue transmission curve can be determined or approximated using the four transmission values. The slope then determines the change in the tissue transmission in a band of a predetermined width (100 nm in this example) around the center of the LED band. We denote the slopes between 627 to 645nm and 645 to 670nm by A and B, respectively. This definition of the slopes is expressed as:

$$A = \frac{FLT(\lambda_2) - FLT(\lambda_1)}{(\lambda_2 - \lambda_1) * (FLT(\lambda_1) + FLT(\lambda_2)) / 2} * 100$$

and



$$B = \frac{FLT(\lambda_3) - FLT(\lambda_2)}{(\lambda_3 - \lambda_2) * (FLT(\lambda_2) + FLT(\lambda_3)) / 2} * 100 ,$$

where  $FLT(\lambda_i)$  is the measured FLT value determined at wavelength  $\lambda_i$ . The estimation of these slopes can be improved by calculating the curvature at the center LED (645nm). This curvature (change of the slope/nm) is

$$5 \quad curv = \frac{B - A}{(\lambda_3 - \lambda_1) / 2} .$$

Finally the expressions are obtained for the slopes  $s$  at the three red wavelengths using  $A$  and  $B$  as parameters:

$$\begin{pmatrix} s_{\lambda_1} \\ s_{\lambda_2} \\ s_{\lambda_3} \\ s_{\lambda_4} \end{pmatrix} = \begin{pmatrix} A - curv * (\lambda_2 - \lambda_1) / 2 \\ (A + B) / 2 \\ B + curv * (\lambda_3 - \lambda_2) / 2 \\ -0.5 \end{pmatrix} \quad (8),$$

10 where the slope at the IR wavelength has been estimated to be constant as it cannot be determined by the other LEDs. If we had had another LED, at about 800-1000nm range, for example, it could have been used for the estimation of the IR slope. Because the extinction curves are very flat at 870nm and the transmission is usually rather high, the tissue prefilter cannot alter the effective extinction coefficient from its nominal value significantly. The  
15 approximation of a constant transmission slope is thus considered sufficient.

In principle these slopes could be inserted in Eq. 4 in order to integrate the new true values for the extinction coefficients. However, this is impractical to do in real-time, so a simpler algorithm is presented below. We first calculate off-line the relative changes of extinction coefficients for each analyte of the  
20 system using Eq. 4 and assuming that the value of the slope equals a predetermined value, which is one in this example. This calculation results in a shift matrix of Eq. 9:

$$Tissue_{SHIFT}^{SLOPE=1} = \begin{pmatrix} 0.975 & 0.942 & 0.940 & 0.999 \\ 0.984 & 0.966 & 0.920 & 0.916 \\ 0.963 & 0.986 & 0.896 & 0.789 \\ 1.01 & 1.03 & 0.903 & 1.05 \end{pmatrix} \quad (9),$$

25 where the effective extinction of HbO<sub>2</sub> at 645nm is 0.966 times the original value, and the effective extinction of HbCO at 670nm is 0.896 times the original value, for example. Thus, the matrix of Eq. 9 defines the relative changes caused by the tissue, assuming that the slope of the tissue

transmission curve equals one. During in-vivo measurement, the slope is estimated using the DC values. The ratio of the slopes then indicates the relative change of a coefficient. In other words, if the relative change determined off-line for a slope value of  $X1$  is  $r$ , the relative change for a measured slope value of  $X2$  is then  $r_X(X2/X1)$ . The relative changes are different for different analytes since the extinction coefficients of the different analytes behave differently as a function of wavelength.

In real-time the effective extinction coefficients can thus be calculated as follows:

$$E^{Eff} = E_{kl}^0 \otimes (1 + S \bullet (Tissue_{SHIFT}^{SLOPE=1} - 1)) \quad (10),$$

where  $S$  denotes the column array in Eq. 8 and the matrix multiplications are performed element by element ( $\otimes$ ) or element by row ( $\bullet$ ), respectively.

If the LEDs are not driven at the nominal drive currents, their effective wavelength may also be shifted by the temperature change at the LED p-n junction. The wavelength shift induced by temperature is typically about 0.1-0.2 nm/°C which is significant if the drive currents are high, as is usually the case at wavelengths shorter than 660 nm. Thus, the extinction matrix of Eq. 10 must also be compensated for in varying LED drive conditions.

There are many ways to find out the temperature of the LED p-n junction. One alternative is to add a temperature sensor on the LED substrate and use the reading of the sensor for the compensation of all LED emission wavelengths. Though the junction temperature follows the substrate temperature according to some empirical heat conduction model, the method may be unreliable because the LED chip contact to the substrate and the internal heat conductivity may vary considerably. A better way is therefore to determine the junction temperature directly from the forward voltage drop of the LED junction. The junction has typical diode characteristics, which can be determined off-line for each LED separately after assembling the LEDs on the substrate. It is even possible to measure, with an optical spectrometer the shift of the emission as a function of the LED forward voltage. Relating the wavelength shift to the forward voltage assumes that the forward voltage is measured during the operation of the pulse oximeter. The circuit board of the pulse oximeter should thus preferably have means for performing the forward voltage measurement. But if it does not, the LED emission shift can be calibrated against the temperature sensor at the substrate. The LED manufacturer specifications for the temperature shift can then be used to

calculate the corresponding wavelength shift. Still another practical compensation for the emitter temperature changes is to map empirically the relationship of the emitter drive current to the observed wavelength shift and to use this information to adjust the on-line extinction coefficients for the sensor.

- 5 A method for temperature compensation of the LED emission is now presented, assuming that the LED forward voltage is measured on the circuit board. The wavelength shifts can then be calculated as follows

$$\begin{pmatrix} \Delta\lambda_1 \\ \Delta\lambda_2 \\ \Delta\lambda_3 \\ \Delta\lambda_4 \end{pmatrix} = \begin{pmatrix} k_1 \\ k_2 \\ k_3 \\ k_4 \end{pmatrix} \bullet \begin{pmatrix} \Delta V_1 \\ \Delta V_2 \\ \Delta V_3 \\ \Delta V_4 \end{pmatrix} \quad (11),$$

- 10 where the shift coefficients  $k_i$  are values determined empirically in advance and  $\Delta V_i$  are the measured changes of the forward voltage drops. For the 627-645-670-870nm LEDs of the sensor, the k-values are 0.06 nm/mV, 0.06nm/mV, 0.09nm/mV, and 0.1nm/mV, respectively.

- 15 As in the compensation discussed above relating to tissue filtering, it is practical to first calculate the change in the extinction coefficients off-line for a certain fixed wavelength shift. In this example the relative changes of the extinction coefficients are calculated, as in Eq. 9, for a 5 nm wavelength shift for each of the four hemoglobin derivatives. The following shift matrix is then obtained:

$$Temp_{SHIFT}^{\Delta\lambda=5nm} = \begin{pmatrix} 0.919 & 0.820 & 0.798 & 0.974 \\ 0.963 & 0.926 & 0.823 & 0.794 \\ 0.941 & 0.983 & 0.855 & 0.725 \\ 1.0 & 1.01 & 0.963 & 1.02 \end{pmatrix} \quad (12).$$

- 20 During in-vivo measurement the relative changes are then calculated based on the measured wavelength shift. The ratio of the wavelength shifts then indicates the relative change of a coefficient caused by temperature. In other words, if the relative change calculated for a wavelength shift of Y1 is  $r$ , the relative change for the measured (on-line) wavelength shift of Y2 is  $r \times (Y2/Y1)$ . The relative changes are different for different analytes, since the extinction coefficients of the different analytes behave differently as a function of wavelength.

The temperature compensated extinction coefficients are thus:

$$E_{TEMP}^{EFF} = E_{kl}^0 \otimes (1 + (\Delta\lambda / 5nm) \bullet (Temp_{SHIFT}^{\Delta\lambda=5nm} - 1)) \quad (13),$$

where  $\Delta\lambda$  is the array in Eq. 11. As mentioned earlier,  $\Delta\lambda$  can also be estimated by reading the temperature indicated by the temperature sensor on the LED substrate or by measuring the LED drive current and using the mapping of the current to the wavelength shift.

- 5 The compensation of the variability causing wavelength shift type interference can now be summed up as follows:

$$E^{Eff} = E_{kl}^0 \otimes (1 + S \bullet (Tissue_{SHIFT}^{SLOPE=1} - 1)) \otimes (1 + (\Delta\lambda / 5nm) \bullet (Temp_{SHIFT}^{\Delta\lambda=5nm} - 1)) \quad (14).$$

As mentioned above, the practical compensations can be divided into steps performed off-line prior to the actual measurements, for example in the factory at the manufacturing stage of the pulse oximeter sensor, and to steps performed on-line, i.e. in connection with the actual in-vivo measurements. Using the above practical example, the off-line compensation steps are as follows.

1. The spectral characteristics of the emitter/detector system is established. In other words, the LEDs are characterized for their light emission (the emission as a function of wavelength) and the detector for its spectral sensitivity. This step thus includes determination of the characteristics of the curve shown in FIG. 9, i.e. the received intensity as a function of wavelength (at least around the wavelengths used). The light transmission from the light emitter to the light detector is measured without human tissue, i.e. the CTR is determined in the sensor in a fixed setup mimicking the actual use of the sensor. For clip-type sensors this is usually the Probe Off position of the sensor. The step also includes determination of the center wavelength of each LED.

2. The effective extinction coefficients are determined without the tissue term. Thus in this step Eq. 4 is used without the tissue term ( $tissue(\lambda)$ ) to form a matrix  $E_{kl}^0$  according to Eq. 5.

3. The relative changes in the effective extinction coefficients due to tissue filter effect are determined. Here Eq. 4 is used assuming that the slope of  $tissue(\lambda)$  equals a predetermined fixed value at each wavelength. In other words, the shift matrix of Eq. 9 is determined.

4. The relative changes in the effective extinction coefficients due to wavelength shift caused by changes in temperature are determined. In other words, the matrix of Eq. 12 is determined, which indicates the relative changes for a wavelength shift of a predetermined value.

5. If the LED forward voltage method is used, the LED forward voltages are characterized at a typical drive current for small ambient

temperature changes. The temperature coefficients  $k_i$  in Eq. 11 and the nominal forward voltage drops at nominal temperature are determined for each LED.

6. The information obtained is saved in a memory unit in the sensor or in the control unit, or the corresponding information is otherwise made available to the pulse oximeter, for example, by using codes, such as sensor identification numbers, which indicate the values of the information.

As is obvious from the above, steps 1 to 6 include performing initial measurements for the compensation, said measurements to include measuring the light transmission of the apparatus, establishing nominal DC transmission characteristics of the apparatus on the basis of the measurements, and for subsequent in-vivo measurements storing reference data which indicate the transmission characteristics established.

- Thus in the off-line measurements, the value of Eq. 4 is determined using nominal values, and the nominal extinction matrix is formed. The apparatus is also provided with the data needed in the subsequent on-line compensation steps for calculating the changes in the factors included in Eq. 4. In the on-line steps the said changes are determined and a new extinction matrix is formed, whereby the new extinction values are such that the external effects are compensated for.

To sum up, after the off-line steps the pulse oximeter stores the matrices according to equations 5, 9, and 12 and the values of the shift coefficients  $k_i$ . In addition to this, the oximeter stores the CTR values and the center wavelengths corresponding to these values.

- Following the above practical algorithm, the on-line compensation steps utilizing the information stored during the off-line steps are as follows:

1. The forward voltage drops ( $\Delta V$ ) are determined for each LED. In this step, the change in the forward voltage relative to the sensor nominal values are calculated, the nominal values having been stored in the sensor memory unit or in the control unit memory. If the pulse oximeter does not have forward voltage measuring means, the temperature of each LED is estimated by reading the temperature indicated by a sensor on the LED substrate, and either the manufacturer specifications or empirical data for corresponding wavelength changes or a look-up table mapping the emitter drive current to the center wavelength shift is used.

2. The effective extinction matrix for the temperature compensation is determined. This is calculated as illustrated in Eq. 11-13, i.e. using the

wavelength shifts determined according to Eq. 11 and calculating the matrix of Eq. 13 using the matrices of Eq. 5 and 12 stored earlier.

3. The DC light transmission for each LED channel (wavelength) is measured, and the value measured is normalized to an emitter current value of 1 mA. The result is the FCTR of the sensor. An estimate for the FLT is then calculated for each emitter (wavelength). In this connection, the FLT values are for calculating the slopes (A and B). In other words, all DC values are normalized in relation to the 1 mA emitter current in order to make all values comparable to one another.

5. The effective Lambert-Beer extinction coefficients are determined using Eq. 14.

6. The extinction coefficients obtained in step 5 are used for solving the analyte concentrations in the Lambert-Beer model.

During the above on-line steps 1-6 in-vivo measurements are performed, wherein the DC component of the radiation emitted through the tissue and received by the detector is measured, tissue-induced changes in the transmission characteristics are determined based on the in-vivo DC component and the transmission characteristics stored, and on the basis of the tissue-induced changes the subject-specific variation in the in-vivo measurement is compensated for.

The above off-line and on-line steps compensate for the non-ideal characteristics of the broadband emitters or for external effects on the light source emission spectra. However, they do not compensate for the variation in the absorption and scattering interplay in the tissue, i.e. the internal effects, which equally influence a single line laser emitter and a broadband LED emitter. Lasers also show shifts in the emission line wavelength as a function of the temperature. Therefore, the lasers are compensated for the temperature and internal tissue effects, but not for the pre-filter tissue-induced spectral shifts.

The compensation step attending to the variations in the internal effects is illustrated in the flow diagram of FIG. 10. Initial measurements are again made in order to determine the nominal DC transmission characteristics of the pulse oximeter (steps 101 and 102). As a result of these steps, reference data are created and stored, and the data indicating the transmission characteristics are determined (step 103). On the basis of the data and the in-vivo measurements made, the tissue-induced change in the average transformation is then determined in connection with the in-vivo measurements

(step **104**). The new transformation is determined, and finally the concentrations of the analytes are solved in the Lambert-Beer model (step **105**). Thus in this embodiment the average transformation stored in the pulse oximeter is changed in a subject-specific manner in order to improve the accuracy of the pulse oximeter. As shown below, in this case during the off-line steps **101-103** partly different data is to be created and stored as compared to the first compensation step, since in this case the DC light transmission characteristics are used to create a subject-specific transformation.

A practical implementation of the second compensating step is now discussed by introducing a new variable called “path length multiplier”, since this will provide an easy way of understanding the technique in accordance with the invention.

As mentioned above, the purpose of the invention is to improve the accuracy of a pulse oximeter in situations in which the blood volume, the red blood cell density or the hematocrit, the total hemoglobin (g/dl), the division between the arterial and venous blood compartment volumes, and the arterial-venous saturation difference vary and produce human variability, which worsens the accuracy of the SpO<sub>2</sub> measurement. It is also the purpose of the invention to compensate for the effect of skin pigmentation (dark skin), which in part can be considered to belong to the tissue prefilter category of compensations, but which also influences via modifying the path length multiplier. This modification is especially important for SpO<sub>2</sub> ear sensors, which are attached to a very thin tissue part (of about the same thickness corresponding to the diffusion constant in human tissue).

The interdependence of the above-described transformation and the path length multiplier is first illustrated by considering the photon path lengths through a single layer of artery blood and examining how the scattering and absorption affect it. It is postulated here that multiple scattering effectively increases the photon path length through the artery and that the absorption of the surrounding tissue effectively decreases it. In this way the artery and tissue are in interaction with each other. To derive a mathematical formulation of this relationship, the known Kubelka-Munk two-flux model can be used. This model defines an absorption probability  $K$  as follows:

$$K = \left\langle \frac{dl}{dz} \right\rangle * \Sigma_a \quad (15),$$

where  $\Sigma_a$  is the macroscopic absorption cross-section of the media and  $dl$  is the true average photon path length through the scattering and absorbing

medium of infinitesimal layer thickness  $dz$ . The term  $\langle dl/dz \rangle = K/\Sigma_a$  is a path length multiplier ( $plm$ ) which enhances the arterial blood absorption from that of the Lambert-Beer non-scatter value because of the multiple scattering in the surrounding medium.

- 5        The idea of the path length multiplier is applied to the Lambert-Beer formulation of 2- $\lambda$  pulse oximetry. The ratio of the change in absorption at the two probe wavelengths is defined as:

$$\frac{dA_k}{dA_l} = N_{kl}^{in-vivo} = \frac{\mu_a^k * d_k}{\mu_a^l * d_l} \quad (16),$$

- 10        where  $\mu_a^i$  is the arterial (non-scatter) absorption coefficient at wavelength  $i$  and  $d_i$  is the effective true optical path length. The transformation is defined by substituting equation 15 with  $dl = d_i$  in equation 16:

$$N_{kl}^{in-vivo} = \frac{\mu_a^k * \left( \frac{K}{\Sigma_a} \right)_k * dz}{\mu_a^l * \left( \frac{K}{\Sigma_a} \right)_l * dz} = \left( \frac{K}{\Sigma_a} \right)_k * N_{kl}^{ideal} \quad (17),$$

- 15        where the ideal Lambert-Beer model is used for  $N_{kl}^{ideal} \equiv \mu_a^k / \mu_a^l$ , and where the layer thickness  $dz$  is the same for all wavelengths ( $k, l$ ). Equation 17 now represents the transformation  $(g_{kl})^{-1}$  from  $N_{kl}^{in-vivo}$ , i.e. from the measured value, to  $N_{kl}^{L-B}$ , which is the ratio of differential absorptions that would be measured if the measurement system were the ideal cuvette system of the Lambert-Beer model. For the transformation the equation below is obtained:

$$g_{kl} = \frac{\left( \frac{K}{\Sigma_a} \right)_k}{\left( \frac{K}{\Sigma_a} \right)_l} = \frac{plm_k}{plm_l} \quad (18).$$

- 20        Thus the transforming quantity is a ratio of path length multipliers measured at two different wavelengths ( $k$  and  $l$ ). The dependence of the function  $g_{kl}$  thus refers to the absorption density of the scattering tissue in the surrounding of the infinitesimal arterial layer  $dz$  including the layer itself. This essentially means that the transformation does not require knowledge of the
- 25        analyte composition in the arterial blood, but refers rather to the macroscopic light absorption, i.e. transmission through the tissue part under the sensor. That



is in the language of pulse oximetry the DC component of the light transmission. This is utilized in the compensation of the invention.

Modifying Eq.1 and leaving the attenuation of the probe design factors (i.e. CTR values) out of consideration, the relationship of the DC light transmission through the tissue and the path length multiplier can be presented as follows:

$$I_{out} = I_{in} e^{-\varepsilon DC} = I_{in} e^{-\varepsilon plm D1 C} \quad (19),$$

where  $D$  is the actual path length through the sample,  $D1$  is the shortest path length through the sample (i.e. the thickness of the sample), and  $\varepsilon$  is the ideal extinction coefficient of the analyte. Here the  $I_{out}/I_{in}$  can be associated with the FLT at the wavelength in question.  $Plm$  thus describes the internal attenuation factors in the tissue and, in particular, the enhancement of the absorbancy relative to the ideal cuvette absorption.

In nominal conditions, the path length multiplier has a certain nominal value  $plm^0$  (where the superscript '0' refers to the nominal value). This nominal value can be determined in the factory at the manufacturing stage of the pulse oximeter. When the DC component is measured again in connection with in-vivo measurement, the change in the  $plm$  from the nominal value can be used to determine the change in the average transformation.

The term  $\alpha_{int}$  in Eq. 6 can be expressed with the help of the path length multiplier in the Lambert-Beer model as

$$\alpha_{int} = plm * \Sigma_a,$$

where  $\Sigma_a$  accounts for all internal absorption sources and is defined in the non-scatter Lambert-Beer domain. The FLT at wavelength  $k$  can then be written as follows:

$$FLT_k = \exp(-\alpha_{int} * d) = \exp(-plm * \Sigma_a * d) \quad (20).$$

We then ratio the logarithms of the FLTs at two wavelengths  $k$  and  $l$ , which results in Eq. 21:

$$\frac{\log(FLT_k)}{\log(FLT_l)} = \frac{plm_k * \Sigma_a^k}{plm_l * \Sigma_a^l} = g_{kl} * \frac{f_a * \mu_a^k + f_v * \mu_v^k}{f_a * \mu_a^l + f_v * \mu_v^l} = g_{kl} * \frac{f_a(\mu_a^k - \mu_v^k) + \mu_v^k}{f_a(\mu_a^l - \mu_v^l) + \mu_v^l} \quad (21),$$

where  $g_{kl} = plm_k / plm_l$  is the transformation between the Lambert-Beer and in-vivo modulation ratios according to Eq. 18, and in which the internal absorbing tissue compartments are venous and arterial blood with volume fractions  $f_v$  and  $f_a$  and with absorption coefficients  $\mu_{lv}$  and  $\mu_{la}$  determined in the Lambert-Beer domain, respectively. In the last expression we have used for the

venous volume fraction the relationship  $f_v = 1 - f_a$ . As the arterial volume fraction is always smaller than the venous volume fraction and as the arterial-venous absorption difference is always smaller than the venous absorption, the dominating factor in the last term is  $\mu_v^k / \mu_v^l$ , i.e. the venous saturation SvO2.

- 5 Thus the changes in the FLT and SvO2 from their nominal values provide the compensation needed for estimating the correct transformation function  $g_{kl}$ . We can then finally write for the relative change of the transformation function  $g_{kl}$ :

$$\frac{g_{kl}}{g_{kl}^0} = \frac{\frac{\log(FLT_k)}{\log(FLT_l)} \frac{\log(FLT_k)^0}{\log(FLT_l)^0}}{F_{kl}(SvO2, SaO2, f_a) / F_{kl}(SvO2, SaO2, f_a)^0} \quad (22),$$

- 10 where the function  $F_{kl}$  represents the ratio term  $\frac{f_a(\mu_a^k - \mu_v^k) + \mu_v^k}{f_a(\mu_a^l - \mu_v^l) + \mu_v^l}$  in

Eq. 21 and the superscript 0 represents the nominal values of the nominal calibration function  $g_{kl}^0$ , which is on average true for a large patient population.

- In fact, the  $\log(FLT)$  and the  $F_{kl}$  compensation terms account for quite different human variability factors in the tissue: whereas  $F_{kl}$  mainly tracks the changes of the arterial venous saturation difference, in particular SvO2, the  $\log(FLT)$  reflects the changes in the total absorption of the tissue, i.e. in the total blood volume and the total hemoglobin or hematocrit, which are not seen in  $F_{kl}$  at all. In practice, the largest corrections to the transformation function are due to the  $\log(FLT)$  and  $F_{kl}$  is less important.

- 20 The FLT in Eq. 22 is easily obtained at the two wavelengths  $k$  and  $l$ , as has been described earlier in Eq. 7. The function  $F_{kl}$  represents the ratio of the absorption coefficients (in the Lambert-Beer non-scatter model) of the whole tissue at these same two wavelengths, i.e. it represents the internal color of the tissue. This internal absorption ratio can be measured by examining the low frequency baseline fluctuations of the plethysmographic wave signal. These fluctuations are caused by the low frequency changes (usually of respiration origin) in the blood volume or in the blood volume distribution of the tissue. Similarly, since the arterial color (= R-ratio) is defined as the ratio of the arterial absorption coefficients, function  $F_{kl}$  can be calculated as:

30 
$$F_{kl} = \frac{f_a(\mu_a^k - \mu_v^k) + \mu_v^k}{f_a(\mu_a^l - \mu_v^l) + \mu_v^l} = g_{kl}^{-1} \left( \frac{(AC/DC)_k}{(AC/DC)_l} \right) = g_{kl}^{-1} (N_{kl}^{baseline}) \quad (23),$$

where  $AC$  is the amplitude (or the instantaneous slope) of the low frequency baseline fluctuation, instead of the heart pulse amplitude of the plethysmographic wave, and  $DC$  is the DC light transmission at that particular wavelength. Because the effective tissue color is mainly determined by the venous blood, function  $F_{kl}$  can be approximated as the arterial modulation ratio calculated for the venous saturation, which is usually about  $SaO_2-10\%$  i.e.  $F_{kl}=R(SvO_2=SaO_2-10\%)$ .

If the venous saturation is determined by venous blood samples and the arterial saturation by the arterial blood samples, the function  $F_{kl}$  can be calculated using the real blood values (with the assumption that the corresponding blood compartment volumes are  $f_a=0.25$  and  $f_v=0.75$ ).

The second compensation step for human variability can then be performed in the following way:

1. The nominal light transmission through a finger or ear lobe or its approximation at the distinct nominal wavelengths of the sensor, is determined for a population of patients or volunteers or even for only one single volunteer, on whom the nominal calibration (the original standard pulse oximeter calibration or the actual multi wavelength measurements using this calibration directly or indirectly) was performed; this gives  $\frac{\log(FLT_k)^0}{\log(FLT_l)^0}$  as a function of  $N_{kl}^{in-vivo}$  at these wavelengths ( $k, l$ ), i.e. as a function of the correct  $SpO_2$  in practice. Thus, the  $FLT^0$  values are obtained for each wavelength. This information is stored in the sensor memory unit (or in the control unit).

2. Function  $F^0$  can be determined in two alternative ways:

- 2.1 In the above measurement using the baseline fluctuations,  $N_{kl}^{baseline}$  is calculated and transformed to the Lambert-Beer model by the nominal transformation  $(g_{kl}^0)^{-1}$  for the population of patients or volunteers or even for only one single volunteer, on whom the nominal calibration (the original standard pulse oximeter calibration or the actual multi-wavelength measurements using this calibration directly or indirectly) was performed; this is tabulated as a function of  $N_{kl}^{in-vivo}$  at these wavelengths ( $k, l$ ). These data are stored in the sensor memory unit (or in the control unit).

- 2.2. In the above measurement venous and arterial blood samples are taken from a position close to the sensor site and analyzed for RHb, HbO<sub>2</sub>, HbCO, and metHb. The absorption coefficients  $\mu_a$  and  $\mu_v$  are then calculated using the measured analyte fractions. The arterial volume fraction  $f_a$  is then estimated. Usually it is sufficient to approximate that  $f_a$  is equal to 0.25. The

functions  $F_{kl}$  are calculated in the Lambert-Beer Model using the venous and arterial volume fractions and

$$\mu_{a,v} = RHb * \mu_{a,v}^{RHb} + HbO2 * \mu_{a,v}^{HbO2} + HbCO * \mu_{a,v}^{HbCO} + metHb * \mu_{a,v}^{metHb}.$$

3. As a result of steps 1 and 2, we have an off-line determined look-up  
5 table for each wavelength pair, in which the following nominal information is tabulated:

k	l	$N_{kl}^{in-vivo}$	$g_{kl}^0$	$\text{Log}(FLT_k)/\text{log}(FLT_l)$	$g_{kl}^{-1} \times (N_{kl}^{baseline})$	$F_{kl}$

Table 1.

- and where also the nominal transformation  $g^0$  is presented and  
10  $N_{kl}^{baseline} \cong g_{kl}^0 \times (F_{kl})$ . Only one of the two last columns is necessary, depending on the way the values of function  $F$  are determined. It is also to be noted that the ratio is not necessary in column 5, but that it is enough to store the FLT values from which the ratio of their logarithms can be calculated.

- As is obvious from the above, steps 1-3 again include measuring the  
15 DC light transmission of the apparatus, establishing nominal DC transmission characteristics for the apparatus on the basis of the measurements, and storing reference data for subsequent in-vivo measurements, the data indicating the transmission characteristics established.

- After the above off-line steps, which can be performed at the  
20 manufacturing stage of the pulse oximeter, the following steps are performed in connection with the on-line (i.e. in-vivo) measurement.

4. The actual spectral shape of the light transmission through a finger or ear lobe at the distinct wavelengths of the sensor is determined, and  
 $\frac{\log(FLT_k)}{\log(FLT_l)}$  is calculated for each wavelength pair similarly as in the off-line  
25 stage.

5. In Table 1 in the row where the measured modulation ratio at the wavelengths  $k$  and  $l$  equals  $N_{kl}^{in-vivo}$ , the nominal  $\frac{\log(FLT_k)^0}{\log(FLT_l)^0}$  is read. The correction factor  $\frac{\log(FLT_k)}{\log(FLT_l)} \bigg/ \frac{\log(FLT_k)^0}{\log(FLT_l)^0}$  is then calculated.

6. In Table 1 in the row, where the measured modulation ratio at the wavelengths  $k$  and  $l$  equals  $N_{kl}^{\text{in-vivo}}$ , the nominal  $F_{kl}^0$  is read from either of the last two columns.

7. The correction factor  $F_{kl}/F_{kl}^0$  is determined in one of the following two ways:

7.1 Column  $g^{-1} \times (N^{\text{baseline}})$ : By using the baseline fluctuations of the measured plethysmographic wave and by using Eq. 23 with  $g^{-1}$ , the function  $F_{kl}$  is determined. The correction factor  $F_{kl}/F_{kl}^0$  is then calculated. For determining  $N^{\text{baseline}}$  and its changes, the amplitudes of the signal can be used, as is normally done for a modulation ratio  $N$ .

7.2 Column  $F_{kl}$ : The blood analytes RHb, HbO<sub>2</sub>, HbCO, and metHb are solved in the Lambert-Beer model using the nominal transformation  $g^0$ . This is the first approximation for the analytes. The absorption coefficients  $\mu_a$  and  $\mu_v$  are calculated using the measured analyte fractions in the arterial absorption and approximating the absorption coefficient in the venous blood by using the measured dyshemoglobin fractions and setting HbO<sub>2</sub><sup>vena</sup>=HbO<sub>2</sub>-10% and RHb<sup>vena</sup>=RHb+10%, where  $f_a = 0.25$  is assumed. The functions  $F_{kl}$  are calculated in the Lambert-Beer Model using the equation:

$$\mu_{a,v} = RHb * \mu_{a,v}^{RHb} + HbO_2 * \mu_{a,v}^{HbO_2} + HbCO * \mu_{a,v}^{HbCO} + metHb * \mu_{a,v}^{metHb}.$$

A more accurate estimate of  $F_{kl}$  can be obtained by iteration of new analyte fractions for the new corrected transformation after step 8.

8. A new transformation  $g_{kl}$  is calculated using Eq. 22, and the new transformation is used for solving the analyte concentrations in the Lambert-Beer model.

The pre-calculated data utilized by the pulse oximeter can be stored in the sensor part of the pulse oximeter, whereby the same sensor can be attached to different pulse oximeter housings. FIG. 12 illustrates the general structure of a sensor according to the invention, the detailed configuration of the sensor being dependent on which information is stored in the sensor and which in the signal processing part, and also on the amount of the calculation appropriate in the signal processing part.

Nevertheless, a sensor according to the invention includes the light sources (10a-10c) and the photo detector (12), the light sources being adapted to emit at two or more wavelengths. In addition, the sensor includes a data storage unit M2 for storing the data on the basis of which the signal processing part can perform the above-described calibration. The information necessary for the above compensations is shown in the figure. For the first compensation

process the pulse oximeter needs the k-values, the above-mentioned three matrices, i.e. the nominal extinction matrix (Eq. 5), the shift matrices (Eqs. 9 and 12), and the CTR/wavelength pairs. For the second compensation process, in turn, the pulse oximeter needs the information stored in Table 1 and the CTR/wavelength pairs. As mentioned above, at least part of this data determined prior to the use of the device for in-vivo measurements can also be stored in the control unit part of the pulse oximeter. The apparatus further preferably includes means 120 for measuring the forward voltage of the p-n junction of each LED, as discussed above.

As can be seen from the above, the method of the invention is based on the DC transmission of light. By means of the off-line DC measurements, reference data is first created. During subsequent in-vivo DC measurements, the reference data is then utilized to filter out human variability from the in-vivo measurement.

Although the method in accordance with the invention has been discussed in connection with a four wavelength pulse oximeter, it can also be employed in a basic two wavelength pulse oximeter. However, the method is more beneficial in a multi-wavelength pulse oximeter where the number of analytes to be measured is greater than two.

In the case of a two-wavelength pulse oximeter, the simplest way to apply the compensation is first to formulate the calibration of the two-wavelength oximeter as a first step using only one transformation function  $g^{-1}$  (e.g. at wavelengths 660nm and 900 nm) and a second step using a two-times-two extinction matrix  $\epsilon$  for these wavelengths and for the two analytes RHb and HbO<sub>2</sub>. The compensation procedures are then identical to the ones presented in the above multi-wavelength method. If the calibration of the two-wavelength pulse oximeter is done in the normal way using a direct mapping of the in-vivo measured R-ratio (=  $N_{660-900}$ ) to the SpO<sub>2</sub> percentage, the compensation steps could for example, be as follows: The wavelength shifts from the nominal LED center wavelength values to a change in the SpO<sub>2</sub> value can first be coded. The wavelength shifts are determined for the temperature component as described in the above multi-wavelength method and for the tissue component by mapping at the two wavelengths the change in the FLT ratio from its nominal value in the calibration conditions to a change in the SpO<sub>2</sub> value from the nominal calibration SpO<sub>2</sub>. The tissue wavelength shift cannot be estimated as accurately as in the multi-wavelength oximeter, but sufficient compensation to the tissue prefilter can still be obtained and the accuracy of the pulse oximeter

can be improved. The last compensation step also includes the compensation for the internal tissue variability, which is summed with the prefilter effect.

Although the invention has been described above with reference to the examples shown in the appended drawings, it is obvious that the invention is not limited to these, but may be modified by those skilled in the art without departing from the scope and spirit of the invention. For example, instead of transformation, any other quantity by which the pulse oximeter can correct the average calibration known to it can be used to eliminate human variability.

The invention has also been described with reference to pulse oximeters for analytes which are in the blood of a subject. The invention, however, can also be applied at different wavelength ranges, e.g. at around 1.5  $\mu\text{m}$  for glucose, at which similar compensation means are called for. The other substances in the tissue modify the effective extinction of the glucose because they alter the path length multiplier at this wavelength. Similarly, the tissue prefilter and temperature effects are taken into account.